Radioactive Ion Beams





Solar abundances of elements

Solar abundance ($Si^{28} = 10^6$) 10¹²Big Bang fusion reactions neutron reactions 1011 10¹⁰ 10⁹ C12 016 10⁸ Ne²⁰ Fe⁵⁶ 107 10⁶ 105 10000 1000 100 Ba¹³⁸ 10 Pb²⁰⁸ Eisengruppe .1 .01 200 160 180 20 40 60 100 140 0 80 120 Mass number

open questions:

- Why is Fe more common than Au ?
- Why do the heavy elements exist and how are they produced?
- Can we explain the solar abundances of the elements?



Spallation & Projectile Fragmentation Reactions







Some early high-energy proton accelerators:

Facility	Energy	from year
Bevatron (Berkeley)	6 GeV	1954
AGS (Brookhaven)	11 GeV	1960
Fermilab (Chicago)	>300 GeV	1967

They were also used to bombard various stable target materials.

These targets were analyzed with radiochemical methods, i.e. γ -spectroscopy with or without chemical separators

Production cross sections and (some) kinematics for suitable radioactive isotopes





High-energy proton-induced nuclear reactions



Important findings:

Energy-independence of cross sections

✤ Bell-shaped Z-distribution for constant A



Mass yields: exponential slope





Proton- and heavy-ion induced reactions give very similar isotope distribution:

Target fragmentation:	$GeV p + A_{target} \rightarrow A$
Projectile fragmentation:	$\operatorname{GeV}/\operatorname{u}\operatorname{A}_{\operatorname{proj}} + \operatorname{p} \to \operatorname{A}$
are equivalent	





Projectile fragmentation reactions



At GeV energies nucleons can be regarded as a classical particles

- Nucleon-nucleon collisions can be treated classically using measured free nucleon-nucleon cross sections (intra-nuclear cascade).
- In these collisions very *little transfer momentum* is exchanged.
- After the cascade the residual nucleus is *highly excited*.
- Heavy-ion projectiles can be treated as a bag of individual nucleons.

Physical models: Two-step approach

Step 1: Intranuclear-cascade models or Abrasion models Step 2: Evaporation calculation





Projectile fragmentation reactions



X



In-Flight Separation of Radioactive Ion Beams





Fragmentation at Relativistic Energies







Radioactive Ion Beams at GSI





FRagment Separator at GSI





Rare Isotope Selection at FRS: $B\rho$ - ΔE - $B\rho$ Selection





Rare Isotope Selection at FRS: $B\rho$ - ΔE - $B\rho$ Selection





Production, Separation, Identification





Production, Separation, Identification









Production of radioactive ion beams



Random removal of protons and neutrons from heavy target nuclei by energetic light projectiles

Target fragmentation





Random removal of protons and neutrons from heavy projectile in peripheral collisions





ISOLDE at CERN





ISOLDE at CERN



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Production targets

- Over 20 target materials and ionizers, depending on beam of interest
- ➢ U, Ta, Zr, Y, Ti, Si, …
- Target material and transfer tube heated to 1500 – 2000⁰ C
- > Operated by robots due to radiation



converter target



standard target





In the early Copenhagen experiments a ten kilo target consisting of a mixture of baking powder [essentially $(NH_4)_2CO_3$] and uranium oxide was used. Fast neutrons from an internal beryllium target in the cyclotron were used to irradiate the external target, and the radioactive isotopes were produced by fission reactions in the uranium. The radioactive noble gases were then diffused out of the target and swept into the ion source of the isotope separator.





Ionization







Beam extraction and separation

- > All produced ions are extracted by electrostatic field (up to 60 kV)
- > The interesting nuclei are mass selected via magnetic field
 - Lorentz force depends on velocity and mass
 - $m_{\Delta m} < 5000$, so many unwanted isobars also get to experiments





Production, ionization, extraction





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Separation



Magnet separators (General Purpose and High Resolution)





Post-acceleration



3MeV*A beam to experiment





Extracted nuclides





High-Intensity and Energy upgrade of Isolde (HIE-Isolde)



GSİ

An example: Selective production of Astatine

Guinness World Records has dubbed this element the rarest on Earth, stating: "Only around 25g of the element astatine occurring naturally"

- \Rightarrow Ionization potential not experimentally deduced
- \Rightarrow Only two atomic transitions were known







Example: Astatine isotopes

- ✤ How to produce pure beams of At isotopes (all are radioactive)?
 - Use laser to ionize them
 - > Determine for the first time the At ionization potential





S. Rothe et al.; Nature Communications 4 (2013), 1835



Determination of the atomic properties of Astatine



- Determination of ionizing potential
- Identification of new atomic transitions
- Comparison with atomic theory
- Scan of ionizing laser: converging Rydberg levels allow precise determination of the IP
- laser spectroscopy
- Test of atomic theory and quantum chemistry
- Properties of chemical homologue Z = 117
- New beams / exotic decay modes: β -fission
- Potential development of ²¹¹At as a medical radioisotope



The resonance ionization laser ion source (RILIS)





40 MV post-accelerator





40 MV post-accelerator



HIE-ISOLDE has innovated many new ideas, particularly in spacesaving solutions. One way the engineers kept the system compact was to build cryomodules that each contain five cavities, not just one (Image: Maximilien Brice/ CERN)





The new linac had to fit into just 16 m of space. "We had to develop a very compact linac. That's what makes it unique. In other facilities, every cavity has its own cryostat but if we had to do that it would be far too long, so we had to squeeze all of them into one cryomodule. We had to have the solenoids fitted too, they're almost the same length as a cavity, so we had to do lots of design, research and development. The biggest challenge was to design in spaces with clearances of just 1 mm," explains Yacine Kadi, project leader for HIE-ISOLDE. (Image: Maximilien Brice/CERN)

Indian Institute of Technology Ropar



The Miniball Germanium Array



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Coulomb excitation experiments at REX-Isolde











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