Outline: Rutherford scattering

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- 1. history
- 2. impact parameter
- 3. distance of closest approach
- 4. orbital angular momentum
- 5. differential cross section
- 6. nuclear radius



Rutherford scattering discovery of nucleus





Rutherford scattering discovery of nucleus





1909: Rutherford, Geiger and Marsden studied in Manchester the scattering of α -particles on thin gold foils.

Aim: from the angular distribution of the scattered α -particles they wanted to gain information on the structure of the scattering center.

Experimental set-up: Ra-source with $E_{kin}(\alpha) = 4.78$ MeV thin Au-foils (Z = 79, d = 2000 atomic layers) detection of scattered α 's with ZnS scintillator



1908: Nobel price for chemistry 'disintegration of elements'



Ernest Rutherford (1871-1937)



The Scattering of α and β Particles by Matter and the Structure of the Atom

E. Rutherford, F.R.S.* Philosophical Magazine Series 6, vol. 21 May 1911, p. 669-688

669





§ 1. It is well known that the α and the β particles suffer deflexions from their rectilinear paths by encounters with atoms of matter. This scattering is far more marked for the β than for the α particle on account of the much smaller momentum and energy of the former particle. There seems to be no doubt that such swiftly moving particles pass through the atoms in their path, and that the deflexions observed are due to the strong electric field traversed within the atomic system. It has generally been supposed that the scattering of a pencil of α or β rays in passing through a thin plate of matter is the result of a multitude of small scatterings by the atoms of matter traversed. The observations, however, of Geiger and Marsden** on the scattering of α rays indicate that some of the α particles, about 1 in 20,000 were turned through an average angle of 90 degrees in passing though a layer of gold-foil about 0.00004 cm. thick, which was equivalent in stopping-power of the α particles being deflected through 90 degrees is vanishingly small. In addition, it will be seen later that the distribution of the α particles for various angles of large deflexion does not follow the probability law to be expected if such

observation:

event rate ~
$$1/\sin^4\left(\frac{\theta}{2}\right)$$

scattering on a point-like atomic nucleus



Rutherford scattering discovery of nucleus

Kinematic of elastic scattered α -particles \implies energy and momentum conservation

α-particles on electrons (Thomson model)

$$\frac{m_e}{m_\alpha}\approx 10^{-4}$$

max. momentum transfer $\Delta p \sim 10^{-4} \cdot p_i$ only small scattering angles $\theta \sim 0^0$

α-particles on Au-nuclei (Rutherford model)

$$\frac{m_{Au-197}}{m_{\alpha}}\approx 50$$

max. momentum transfer $\Delta p \sim 2 \cdot p_i$ scattering angles up to $\theta_{max} \sim 180^0$ (backscattered α -particles)



$$m_{\alpha} = 4 \ GeV/c^2 \qquad \qquad m_e = 0.511 \ MeV/c^2 \qquad \qquad m_{AU\text{-}197} = 197 \ GeV/c^2$$



In an elastic process

 $a + b \rightarrow a' + b'$

the same particles are present both before and after the scattering, i.e. the initial and the final state are identical (including quantum numbers) up to momenta and energy.

The *target b* remains in its ground state, absorbing merely the recoil momentum and hence changing its kinetic energy.

The scattering angle and the energy of the *projectile a* and the recoil energy and energy of the *target b* are unambiguously correlated.





Rutherford scattering

In the repulsive Coulomb potential $V(r) \sim \frac{z \cdot Z \cdot e^2}{r}$ the α -particle experiences a momentum change $\Delta \vec{q} = \vec{p}_f - \vec{p}_i$ $\vec{p}_i \quad (\pi - \theta)/2$ $\Delta q = 2 \cdot m \cdot v \cdot sin\frac{\theta}{2}$ $\Delta q = \int F_{\Delta q} \cdot dt = \int_{-\infty}^{+\infty} \frac{1}{4\pi\varepsilon_0} \cdot \frac{z \cdot Z \cdot e^2}{r^2} \cdot cos\phi \cdot dt$ angular momentum: $\vec{L} = |\vec{r} \times m \cdot \vec{v}| = m \cdot v \cdot b = m \cdot \frac{d\phi}{dt} \cdot r^2$ $\Delta q = \int_{-(\pi - \theta)/2}^{(\pi - \theta)/2} \frac{1}{4\pi\varepsilon_0} \frac{zZe^2}{v \cdot b} \cdot cos\phi \cdot d\phi = \frac{1}{4\pi\varepsilon_0} \frac{zZe^2}{v \cdot b} \cdot 2 \cdot cos\frac{\theta}{2}$

relation between impact parameter **b** and scattering angle θ :

$$\tan\frac{\theta}{2} = \frac{1}{4\pi\varepsilon_0} \frac{z \cdot Z \cdot e^2}{m \cdot v^2} \cdot \frac{1}{b}$$



Rutherford scattering scattering parameters



impact parameter: $b = a \cdot \cot \frac{\theta_{cm}}{2}$

distance of closest approach:

$$D = a \cdot \left[sin^{-1} \frac{\theta_{cm}}{2} + 1 \right]$$

orbital angular momentum: $\ell = k_{\infty} \cdot b = \eta \cdot \cot \frac{\theta_{cm}}{2}$



$$a = \frac{0.72 \cdot Z_1 Z_2}{T_{lab}} \cdot \frac{A_1 + A_2}{A_2} \quad [fm]$$

asymptotic wave number:

$$k_{\infty} = 0.219 \cdot \frac{A_2}{A_1 + A_2} \cdot \sqrt{A_1 \cdot T_{lab}} \quad [fm^{-1}]$$

Sommerfeld parameter:
$$\eta = k_{\infty} \cdot a = 0.157 \cdot Z_1 Z_2 \cdot \sqrt{\frac{A_1}{T_{lab}}}$$



center of mass system



Cross section

The cross section gives the probability of a reaction between the two colliding particles

Consider an idealized experiment:

- we bombard the target with a monoenergetic beam of point-like particles a with a velocity v_a .
- a thin target of thickness d and a total area A with N_b scattering centers b and with a particle density n_b .
- each target particle has a cross-sectional area σ_b , which we have to find by experiment!



 \rightarrow Some beam particles are scattered by the scattering centers of the target, i.e. they are deflected from their original trajectory. The frequency of this process is a measure of the cross section area of the scattered particles σ_b .

Rutherford cross section

• Particles from the ring defined by the impact parameter b and b+db scatter between angle θ and $\theta+d\theta$





an atomic nucleus exist

$$j \cdot 2\pi \cdot b \cdot db = j \cdot 2\pi \cdot \sin\theta \cdot \frac{d\sigma}{d\Omega}$$
$$\frac{d\sigma}{d\Omega} = \frac{b}{\sin\theta} \left| \frac{db}{d\theta} \right|$$
impact parameter: $b = a \cdot \cot\frac{\theta}{2}$
$$\left| \frac{db}{d\theta} \right| = \frac{a}{2} \cdot \frac{-\sin\frac{\theta}{2} \cdot \sin\frac{\theta}{2} - \cos\frac{\theta}{2} \cdot \cos\frac{\theta}{2}}{\sin^2\frac{\theta}{2}} = \frac{a}{2} \cdot \frac{1}{\sin^2\frac{\theta}{2}}$$
$$\frac{d\sigma}{d\Omega} = a \cdot \frac{\cos\frac{\theta}{2}}{\sin\frac{\theta}{2}} \cdot \frac{1}{2 \cdot \cos\frac{\theta}{2} \cdot \sin\frac{\theta}{2}} \cdot \frac{a}{2 \cdot \sin^2\frac{\theta}{2}}$$
$$\frac{d\sigma}{d\Omega} = \frac{a^2}{4} \cdot \sin^{-4}\frac{\theta}{2}$$

Annular gas-filled parallel-plate avalanche counter (PPAC)



A. Jhingan detector laboratory at IUAC



Indian Institute of Technology Ropar



Experimental set-up at IUAC







Elastic scattering and nuclear radius





Elastic scattering and nuclear radius



Nuclear interaction radius: (distance of closest approach)

$$R_{int} = D = a \cdot \left[sin^{-1} \frac{\theta_{1/4}}{2} + 1 \right]$$

$$R_{int} = C_p + C_t + 4.49 - \frac{C_p + C_t}{6.35} \quad [fm]$$

$$C_i = R_i \cdot (1 - R_i^{-2}) \quad [fm] \qquad R_i = 1.28 \cdot A_i^{1/3} - 0.76 + 0.8 \cdot A_i^{-1/3} \quad [fm]$$



J.R. Birkelund et al., Phys.Rev.C13 (1976), 133





¹³⁶Xe on ²⁰⁸Pb at 700 MeV/u

excitation of giant dipole resonance

 $R_{\rm int} = 15.0 \, fm \rightarrow \mathcal{G}_{1/4} = 5.7 \, mrad$



Protons Neutrons

A.Grünschloß et al., Phys. Rev. C60 051601 (1999)



High-energy Coulomb excitation grazing angle



¹³⁶Xe on ²⁰⁸Pb at 700 MeV/u excitation of giant dipole resonance

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